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CHE105 Intro to Chemistry Lab: Sample Lab Your Instructor's Name: Todd Francis Your Name: Courtney Deist Date: 6/9/18 Short Answer Sample Lab Experiment 1 Lab Results 1. Record the maximum temperature reached by the contents of the Erlenmeyer flask. 25.5 C Data Analysis 1. Did a reaction happen

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in the Erlenmeyer flask?

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completely clueless on how to do this and it's due today. For each of the week old orange juice titrations, calculate and record the following data. If you had to perform the fine titration three times, discard the volume that is different.

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Calculate the final volume of the reaction mixture after the contents of beaker B are added to beaker A. Report your answer in liters. All $\text{S}_2\text{O}_3^{2-}$ is consumed at the end of the reaction. Therefore, the moles of $\text{S}_2\text{O}_3^{2-}$ consumed can be calculated using the equation below. (moles $\text{S}_2\text{O}_3^{2-}$) consumed = $M_{\text{stock}} \times (V_{\text{stock}})$. The I_3^- produced in reaction 1 reacts with $\text{S}_2\text{O}_3^{2-}$ in reaction 2 as ...

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